# Synthesis of a Novel UDP-carbasugar as UDP-galactopyranose Mutase Inhibitor

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**S** Supporting Information



ABSTRACT: The multistep synthesis of a novel UDP-C-cyclohexene, designed as a high energy intermediate analogue of the UDP-galactopyranose mutase (UGM) catalyzed isomerization reaction, is reported. The synthesis of the central carbasugar involved the preparation of a galactitol derivative bearing two olefins necessary for the construction of the cyclohexene ring by a ring-closing metathesis as a key step. Further successive phosphonylation, deprotection, and UMP coupling provided the target molecule. The final molecule was assayed against UGM and compared with UDP-C-Galf, the C-glycosidic UGM substrate analogue.

**B** ecause of the emergence of extremely resistant strains,<br>tuberculosis still threatens the world population.<sup>1</sup> Thus, novel strategies to fight Mycobacterium tuberculosis have recently been developed. $<sup>2</sup>$  Among them, UDP-galactopyranose [m](#page-3-0)utase</sup>  $(UGM)^3$  has been validated as a new target because of its involvement in [th](#page-3-0)e biosynthesis of the mycobacterial cell wall.<sup>4</sup> Indeed, [s](#page-3-0)ome UGM inhibitors have been shown to kill the  $b$ acterium, $\frac{5}{3}$  a result that confirmed that the enzyme is essenti[al](#page-3-0) for the survival of the pathogen. $4$  UGM catalyzes the interconv[ers](#page-3-0)ion of UDP-galactopyranose (UDP-Galp) 1 into UDP-galactofuranose (UDP-Galf) 2 ([Fig](#page-3-0)ure 1), the biosynthetic precursor of all galactofuranose-containing eukaryotic and prokaryotic glycoconjugates.6,7 UGMs fro[m](#page-1-0) major pathogens have been identified,  $6a,8$  but interestingly, this enzyme is absent in mammals, which may favor t[he d](#page-3-0)iscovery of selective therapeutic agents.

Moreover, UGM is a unique flavoenzyme whose structure(s) and mechanism have been extensively studied.<sup>6b,9</sup> Surprisingly, it was discovered that the FAD cofactor plays the role of catalytic nucleophile yielding covalent intermediates 5 [an](#page-3-0)d 6 (Figure 1) after the release of the UDP moiety. $66,9i,10$  Several crystal structures of UGM have been obtained, some of them in t[he](#page-1-0) presence of UDP or UDP-Gal $p.^{9\mathrm{b},11}$  Ho[wever,](#page-3-0) many questions remain unsolved regarding both the mechanism and the binding modes of UGM with its substra[te U](#page-3-0)DP-Galf and the key highenergy intermediates. For instance, the  $S_{N}1/S_{N}2$  nature of the substitutions occurring during this isomerization is still under debate,<sup>6b,9h,i,10,12</sup> with some experiments suggesting a cationic intermediate such as 4 and other data indicating a concerted proces[s.](#page-3-0)<sup>9[e,13](#page-3-0)</sup> [The](#page-3-0)refore, the design of new mechanistic probes that mimic UDP-Galf or the transition states of this enzyme<sup>[9g,h,1](#page-3-0)3,14</sup> may help us to better understand its mechanism and the structural requirements for tight binding to this import[ant thera](#page-3-0)peutic target.

Here we describe the synthesis and biochemical evaluation of a novel UDP-cyclitol 7 designed as a transition-state analogue of the UGM-catalyzed ring contraction. Our multistep synthesis involves the construction of a key cyclohexene scaffold through a ring-closing metathesis reaction.

As highlighted above, the UGM-catalyzed interconversion of UDP-Galp 1 into UDP-Galf 2 involves three important intermediates: the acyclic adduct 5 and the oxycarbeniums 3 and 4. In order to obtain high-affinity mechanistic probes, several groups have designed UDP-galactose analogues mimicking either the charged intermediate  $4^{14a,b,15}$  or UDP coupled to acyclic galactitols to mimic intermediate  $5.^{9h,14gh}$  From these studies it can be concluded that UG[M has](#page-3-0) a much greater affinity for the furanose Galf compared to the pyran[ose. In](#page-3-0) addition, the cationic character of the intermediates is probably not a critical parameter to mimic given the fact that cationic UDP-Galf analogues never displayed strong inhibition profiles.<sup>14a,15</sup> On the other hand, the UDP-galactitol (acyclic form) derivatives such as molecul[es](#page-3-0)  $8^{9\text{h}}$  and  $9^{14\text{h}}$  showed strong inhibition pro[fi](#page-3-0)les against UGM.

We thus [re](#page-3-0)asone[d t](#page-3-0)hat UDP-galactose mimicking both the acyclic intermediate 5 and the galactofuranose substructure of 6 may display potent inhibition profiles and would be interesting to cocrystallize with UGMs. We thus designed UDP-carbasugar 7 as an inhibitor of UGM (Figure 1). Compared to UDP-galactitols 8 and 9, the cyclohexene ring has a similarity to a galactofuranose, thanks in part to the geometr[y o](#page-1-0)f the exocyclic C5−C6 diol, while maintaining some similarity with the acyclic intermediate 5. Moreover, the cyclohexene, with two  $sp<sup>2</sup>$  hybridized carbons, may partially mimic the conformation of the oxycarbenium 4. Globally, cyclohexene 7 gathers some important structural features of the transformation of the galactitol 5 to the furanose 6.

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Figure 1. Mechanism of UGM and design of the new high energy intermediate analogue 7.

Our retrosynthetic analysis of 7 is outlined in Figure 2. The key cyclohexene 10 would be obtained by ring-closing metathesis of



the diene 11 which could possibly be prepared from the protected D-galactonolactone 13 via successive Wittig and Grignard reactions. We anticipated that lactone 13 would be easily accessible in a few steps from D-galactose.

Our first objective was the generation of diene 11, precursor of the cyclohexene 10. The synthesis began with oxidation of Dgalactose in the presence of bromine followed by protection of the resulting D-galactono-1,4-lactone as an acetonide to afford the known lactone  $14^{16}$  which, after benzylation, provided intermediate 13 in 61% yield (Scheme 1). The lactone 13 was then transformed into di[ol](#page-3-0) 15 in four steps. The reduction of the lactone by DIBAL-H was directly followed by a Wittig olefination without purification. A p-methoxybenzyl group was then installed prior to standard dihydroxylation resulting in the formation of epimeric diols 15 (in a ratio 87:13) in 84% yield over four steps after a single purification by silica gel chromatography.

In order to synthesize compound 16, the diol 15 was regioselectively homologated: the silylation of the primary alcohol followed by Dess−Martin oxidation afforded the corresponding ketone. Olefination of the resulting carbonyl group under classical Wittig conditions, and subsequent deprotection of the p-methoxybenzyl ether with DDQ in DCM/H<sub>2</sub>O gave 16 in 68% yield over four steps, with a single purification (Scheme 1). Noteworthy, after the cleavage of the





PMB group, the addition of 4-nitrophenylhydrazine was required to remove the byproduct 4-methoxybenzaldehyde which was found to be inseparable from alcohol 16.

The intermediate 16 was oxidized by Dess−Martin periodinane (DMP) to give the ketone 12, which was then treated by allylmagnesium bromide to provide a mixture of two diastereoisomers 11a and 11b in a ratio 33:67 with 62% yield over two steps (Scheme 2). This stereoselectivity was in accordance with Mulzer's study in which the same ratio was observed for the addition of [a](#page-2-0)llylmagnesium bromide (reagent) on related acyclic ketones bearing two different stereogenic centers on each side.<sup>17</sup> In this study, Mulzer had shown that the choice of protecting groups around the ketone strongly influenced the stere[os](#page-3-0)electivity of the Grignard addition. The absolute configurations in molecules 11a and 11b were at first assigned based on Mulzer's model<sup>17</sup> and were confirmed by NOE experiments performed on the cyclized derivatives 17a and 17b.

The second key step of this synthesis was the generation of the cyclohexene ring by ring-closing metathesis. While this reaction has been already used for a general access to various carbasugars,<sup>18</sup> including cyclohexenes,<sup>19</sup> the synthesis of hindered trisubstituted cyclohexenes remains complex and poorly desc[rib](#page-3-0)ed in the literature. Rec[ent](#page-3-0)ly, Kiessling et al.

#### <span id="page-2-0"></span>Scheme 2. Synthesis of the Cyclohexene Ring



have reported the synthesis of epimers of shikimic acid using this powerful procedure.19a We thus investigated this approach starting from the mixture of epimeric dienes  $11(a,b)$ , and it was found that 0.12 equi[v o](#page-3-0)f Hoveyda−Grubbs (II) catalyst and a portionwise addition were necessary for the reaction to go to completion. After 6 h, cyclohexenes 17a and 17b were isolated in 21% and 63% yields, respectively (Scheme 2).

The structures of the diastereoisomers 17a and 17b were confirmed by NOE experiments performed on product 17b in  $CDCl<sub>3</sub>$  which revealed different correlations between H5 and H3, H4′b and H2, and finally the two correlations H4′a−H3 and H4′a−H5. These results confirmed the absolute configuration of compound 17b at C-4.

With the cyclohexene scaffold in hand, we focused on the synthesis of the target nucleotide-sugar analogue 7. Cleavage of the silyl ether in 17b with TBAF in THF gave the alcohol 18 in 84% yield (Scheme 3). Initial investigations on Michaelis−





Becker reaction between a sulfonate or a bromide, generated from 18, and diethyl phosphite failed under various experimental conditions.<sup>20</sup> Fortunately, the Michaelis−Arbusov reaction was successful, but only when the specific conditions recently reported i[n t](#page-3-0)he literature were applied: $^{21}$  the treatment of 18 with trimethyl phosphite and zinc bromide as Lewis acid under microwave irradiation led to a complete [con](#page-3-0)version after 2 h and

the formation of 10 in 53% isolated yield, along with the expected byproduct  $(Me)P(O)(OMe)$ . Compound 10 was then deprotected efficiently with an excess of TMSBr in  $CH_2Cl_2$ followed by hydrogenolysis catalyzed by  $Pd(OH)_2$ . The desired phosphonate 19 was quantitatively obtained after maximum 20 min of reaction time to avoid the double bond saturation.

The cyclohexene phosphonate 19 was then coupled to an activated form of UMP to generate UDP-carbasugar 7, which was obtained in 14% yield after size-exclusion chromatography and reversed-phase HPLC. As for the synthesis of UDP-galactofuranose  $2,^{22}$  the activation of UMP as a N-methylimidazolium salt gave the best results. The low yield can be explained by the fact that on[ce](#page-3-0) formed, the UDP-carbasugar can decompose into a cyclic phosphonate and UMP.<sup>23</sup> This intramolecular reaction occurs not only during the reaction but also during the purification steps. This side [rea](#page-3-0)ction was also observed for other phosphonate analogues of UDP-galactose 1 and 2.<sup>14e,24</sup>

In order to study the inhibition profile of UDP-cyclohexene 7, we used UDP-C- $\alpha$ -Galf 20<sup>24</sup> (the C-glycosidic anal[ogue o](#page-3-0)f UGM's substrate UDP- $\alpha$ -Galf) and UDP 21 for comparison purposes (Table 1). Indeed, [UD](#page-3-0)P is often used in the literature as



 $a$ Inhibition assay conditions: [Inhibitor] = 1 mM, reduced enzyme  $\text{[UGM}_{\text{Kp}}$  = 12 nM,  $\text{[UDP-Galf]}$  = 105  $\mu$ M,  $\text{[dithionite]}$  = 12.5 mM.<br><sup>b</sup> Fluorescence polarization assay conditions:  $\text{[fluorescent probe]}$  = 15 nM,  $[I] = 10 \mu M$  to 2 mM, nonreduced enzyme  $[UGM_{Kp}] = 500 \text{ nM}$ .

a control inhibitor. $9h$  The comparison with 20 was also useful because the replacement of the exo-anomeric oxygen atom by a methylene group [ma](#page-3-0)y affect the mode of interactions with  $UGM.<sup>11a</sup>$ 

Two complementary enzymatic assays were carried out on UGM [of](#page-3-0) Klebsiella pneumoniae  $(\text{UGM}_{\text{Kp}})$  which is the most studied UGM enzyme for inhibition studies.<sup>9h,14b,15,25</sup> UGM is a flavoenzyme whose cofactor must be reduced to be kinetically competent. Thus, we realized first a competi[tion assay a](#page-3-0)gainst the substrate UDP-Galf  $2$  under reducing conditions.<sup>9h</sup> In this assay, the conversion of 2 into its isomer 3 is monitored by HPLC in presence and absence of inhibitors. The decre[ase](#page-3-0)s in reaction rates are translated into inhibition percentages that have been gathered in Table 1.

From this first series of experiments, we noticed that the carbasugar 7 has an inhibition level intermediate between UDP 21 and UDP-C-Galf 20. This result shows that when UGM is catalytically active, the enzyme has a stronger affinity for the cyclohexene analogue 7 than for the galactofuranose subunit of 20. Therefore, these results indicate that, to mimic high-energy intermediates of the UGM catalyzed reaction, it is appropriate, to design molecules in which the galactose unit is indeed between a furanose state as in  $6$  (Figure 1) and an acyclic state as in 5.

However, to obtain a more complete picture of the binding properties of carbasugar 7, w[e](#page-1-0) also measured the fluorescence

<span id="page-3-0"></span>polarization  $(\text{FP})^{26}$  on UGM<sub>Kp</sub> under nonreducing conditions. This assay is easier to perform, but it only gives an estimation of the affinities of the molecules because the enzyme is not in its catalytically active state. The inhibition profile has been found in the following order:  $UDP > UDP - C-Galf > 7$ , thus showing a reversal of affinities compared to the HPLC assay. The latter result was not surprising since major affinity changes of UDPgalactose analogues toward UGM as a function of the redox state of the enzyme have been observed several times in the literature.<sup>9e,24,25,14i</sup>

In summary, we have explored a new synthetic pathway for the generation of the UDP-carbasugar 7 starting from D-galactose and by using ring closing-metathesis as a key step for the transformation of acyclic galactitol into the cyclohexene ring. When assayed against the reduced enzyme, the final cyclohexene 7 displayed a better affinity than the galactofuranoside 20, which constitutes important information for the future design of transition-state analogues.

## ■ ASSOCIATED CONTENT

#### **S** Supporting Information

Experimental procedures and NMR spectra. This material is available free of charge via the Internet at http://pubs.acs.org.

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The authors declare no competing financial interest.

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